



Pergamon

Tetrahedron Letters 40 (1999) 5183–5186

TETRAHEDRON
LETTERS

A [2+2+1] strategy for the conversion of olefins into cyclopentenones. Ring expansion of 2-*N*-methyl-*N*-tosyl-cyclobutanones

Florence Mahuteau-Betzer and Léon Ghosez *

Laboratoire de Chimie organique de Synthèse, Université catholique de Louvain, Place Louis Pasteur, 1,
B-1348 Louvain-la-Neuve, Belgium

Received 10 April 1999; accepted 11 May 1999

Abstract

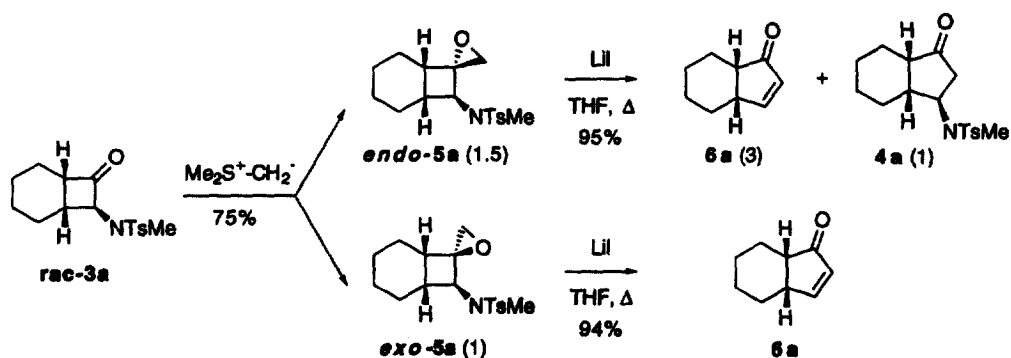
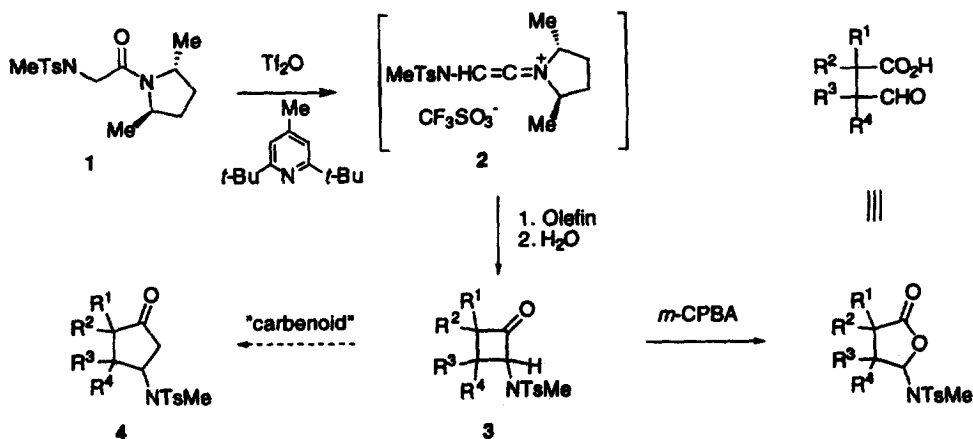
α -*N*-Methyl-*N*-tosylcyclobutanones prepared by asymmetric cycloadditions of the corresponding keteniminium triflate to olefins have been converted into a mixture of epoxides using dimethylsulfonium ylide. Treatment of these epoxides with lithium iodide unexpectedly yielded the corresponding cyclopentenones. This sequence amounts to a [2+2+1] cyclopentannulation of an olefin. © 1999 Elsevier Science Ltd. All rights reserved.

Keywords: cyclobutanones; epoxides; rearrangement; ring expansion; cyclopentenones; cyclopentannulation.

The stereospecific cycloaddition of keteniminium salts to alkenes and alkynes is a very general method of synthesis of four-membered ring ketones.¹ The use of keteniminium salts derived from amides bearing a chiral auxiliary has further extended the utility of the reaction by allowing the preparation of enantiopure cyclobutanones.² The asymmetric cycloaddition of keteniminium salt **2** to an olefin followed by the regioselective oxidation of the resulting cyclobutanone **3** represents one of the few general methods for the enantioselective vicinal acylation of an olefin (Scheme 1).³ More recently we have considered the transformation of cyclobutanones **3** into the corresponding cyclopentanones. If successful, the two-step sequence — cycloaddition followed by ring expansion — would provide a useful method for the asymmetric cyclopentannulation of olefins by a [2+2+1] strategy.

An obvious method for the conversion of **3** to **4** would be the formation of an epoxide using Corey's ylid reagent⁴ followed by a rearrangement catalysed by lithium iodide⁵ or lithium bromide.⁶ The racemic bicyclo[4.2.0]octanone **3a** was readily converted into a 1.5:1 mixture of diastereoisomeric epoxides by reaction with $\text{Me}_2\text{S}^+-\text{CH}_2^-$ generated from trimethylsulfonium tetrafluoroborate and *n*-BuLi (Scheme 2). The two isomeric epoxides were easily separated by flash chromatography.⁷ An X-ray diffraction analysis of *endo*-**5a** confirmed the stereochemical assignments.⁸

* Corresponding author. Fax: +32-10-47.41.68; e-mail: ghosez@chor.ucl.ac.be



Upon treatment with LiI in refluxing THF, the major epoxide *endo*-5a unexpectedly yielded a 3:1 mixture of cyclopentenone 6a and cyclopentanone 4a (Scheme 2, Table 1).⁹ The saturated ketone 4a was the expected product from a regioselective ring-expansion reaction of 5a controlled by the NTsMe group. Cyclopentanone 4a was not the precursor of enone 6a, this was demonstrated by the recovery of unchanged 4a after treatment with LiI for one day in refluxing THF. The rearrangement of *exo*-5a exclusively gave enone 6a. Table 1 shows that both a Lewis acid cation and a nucleophilic anion are necessary for the rearrangement (Table 1, entries 2, 3 versus 1, 4). Also it shows an unusual dependence of the course of the rearrangement on: (1) the nature of the anion (entries 1 and 4); (2) the stereochemistry of the epoxide ring (entries 1 and 4, *endo* versus *exo*). At this stage, we have no good rationale for these unexpected results.

The synthetic value of these observations is illustrated by the efficient two-step conversion of cyclobutanones 3b-d into the corresponding cyclopentenones 6b-d. In these cases both diastereoisomeric epoxides only yielded cyclopentenones 6b-d. The ring expansion step did not require the separation of the isomeric epoxides (Scheme 3). The formation of 6c and 6d from 5c and 5d confirmed that the rearrangement of epoxides 5 took place with the same regioselectivity as that leading to 4.

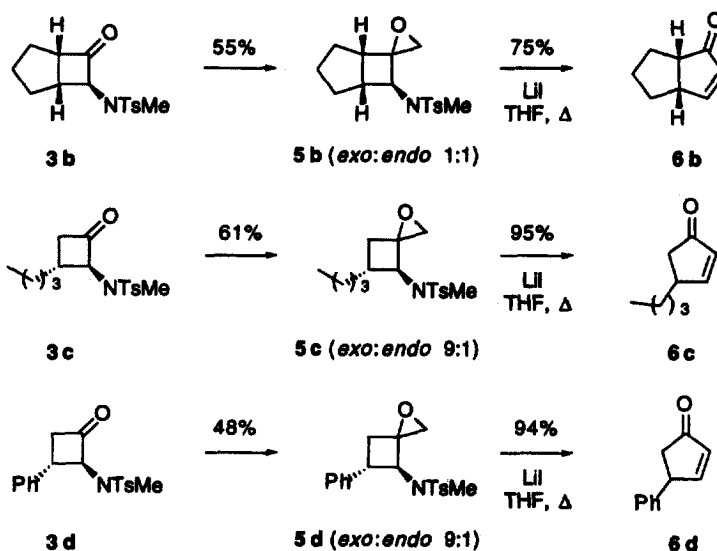
Using the chiral *N*-tosylsarcosinamide 1 for the cycloaddition step,³ we were able to prepare the bicyclic cyclopentenone 6a which showed the same enantiomeric purity as the cyclobutanone 3a (Scheme 4).

This last result indicates that the sequence offers a useful synthetic route for the conversion of olefins into cyclopentenones of high enantiomeric purity. Applications of this synthetic strategy to the

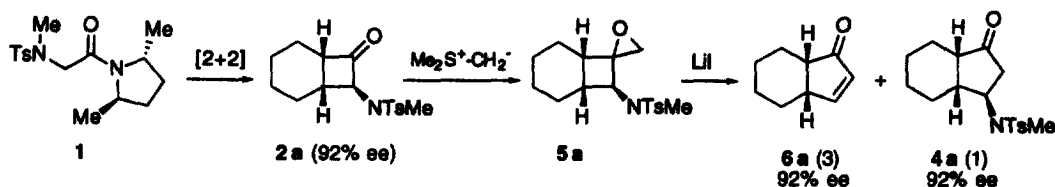
Table 1
Study of the rearrangement of *5a-endo* and *5a-exo* in refluxing THF

Entry	Catalyst (1 equiv.)	<i>endo-5a</i>		<i>exo-5a</i>	
		Yield %	<i>6a</i> : <i>4a</i> ratio ^a	Yield %	<i>6a</i> : <i>4a</i> ratio ^a
1	LiI	95	3 : 1	94	100 : 0
2	LiI, 12-Crown-4	0	-	0	-
3	LINTf ₂	0	-	0	-
4	LiBr	98	0 : 100	95	100 : 0

^a : Determined by NMR on the crude mixture.



Scheme 3.



Scheme 4.

asymmetric synthesis of biologically active products are being studied by our group. The mechanism of the unexpected direct conversion of **5** into **6** is also being investigated.

Acknowledgements

This work was generously supported by the Ministère de l'Éducation et de la Recherche Scientifique de la Communauté française de Belgique (Action concertée 96/01-197).

References

- (a) Schmidt, C.; Falmagne, J. B.; Escudero, J.; Vanlierde, H.; Ghosez, L. *Organic Synthesis* **1990**, *69*, 199–204; (b) Reviews: Ghosez, L. *New Synthetic Methodology and Functionally Interesting Compounds* **1986**, Proceedings of the 3rd International Kyoto Conference on New Aspects of Organic Chemistry, Elsevier Publication, *25*, 99–117; Snider, B. B. *Chem. Rev.* **1988**, *88*, 793–811; Ghosez, L.; Chen, L. Y.; Gobeaux, B.; Houge, C.; Markó, I.; Perry, M.; Saimoto, H. *Strain and its Implication in Organic Chemistry*; de Mejeire, A.; Blechert, S., Eds; Kluwer Academic: Dordrecht, 1989; pp. 235–254; Ghosez, L. *Stereocontrolled Organic Synthesis* Trost, B. M. Ed., Blackwell: London, 1994; pp. 193–233.
- (a) Houge, C.; Frisque-Hesbain, A.-M.; Mockel, A.; Ghosez, L.; Declercq, J.-P.; Germain, G.; Van Meerssche, M. *J. Am. Chem. Soc.* **1982**, *104*, 2920–2921; (b) Saimoto, H.; Houge, C.; Frisque-Hesbain, A.-M.; Mockel, A.; Ghosez, L. *Tetrahedron Lett.* **1983**, *24*, 2251–2254; (c) Chen, L.-Y.; Ghosez, L. *Tetrahedron Lett.* **1990**, *31*, 4467–4470.
- (a) Genicot, C.; Gobeaux, B.; Ghosez, L. *Tetrahedron Lett.* **1991**, *32*, 3827–3830; (b) Genicot, C.; Ghosez, L. *Tetrahedron Lett.* **1992**, *33*, 7357–7360; (c) Ghosez, L.; Genicot, C.; Gouverneur, V. *Pure & Appl. Chem.* **1992**, *64*, 1849–1856.
- (a) Corey, E. J.; Chaykovsky, M. *J. Am. Chem. Soc.* **1962**, *84*, 3782–3783; (b) Review: Trost, B. M.; Melvin Jr., L. S. *Sulfur Ylides: Emerging Synthetic Intermediates*; Academic Press: New York, 1975; 145–151.
- (a) Lriverend, M.-L.; Lriverend, P. C. *R. Acad. Sc. Paris, Série C* **1975**, *t.280*, 791–792; (b) Lriverend, M.-L.; Lriverend, P. *Chem. Ber.* **1976**, *109*, 3492–3495; (c) Morton, D. R.; Brokaw, F. C. *J. Org. Chem.* **1979**, *44*, 2880–2887; (d) Krief, A.; Halazy, S. *J. Chem. Soc., Chem. Commun.* **1982**, 1200–1201; (e) Halazy, S.; Zutterman, F.; Krief, A. *Tetrahedron Lett.* **1982**, *23*, 4385–4388; (f) Newton, R. F.; Wadsworth, A. H. *J. Chem. Soc., Perkin Trans. 1* **1982**, 823–830; (g) Lriverend, M.-L.; Vazeux, M. *J. Chem. Soc., Chem. Commun.* **1982**, 866–867; (h) Wenkert, E.; Arrhenius, T. S. *J. Am. Chem. Soc.* **1983**, *105*, 2030–2033; (i) Hart, T. W.; Comte, M.-T. *Tetrahedron Lett.* **1985**, *26*, 2713–2716.
- (a) Trost, B. M.; Latimer, L. H. *J. Org. Chem.* **1978**, *43*, 1031–1040; (b) Tobe, Y.; Yamashita, S.; Yamashita, T.; Kakiuchi, K.; Odaira, Y. *J. Chem. Soc., Chem. Commun.* **1984**, 1259–1260; (c) Tobe, Y.; Yamashita, T.; Kakiuchi, K.; Odaira, Y. *J. Chem. Soc., Chem. Commun.* **1985**, 898–899; (d) Pirrung, M. C.; Thomson, S. A. *J. Org. Chem.* **1988**, *53*, 227–230.
- General procedure for epoxidation of **3**: a 2.5 M solution of *n*-BuLi in hexane (1.2 equiv.) and a 0.05 M solution of trimethylsulfonium tetrafluoroborate (1 equiv.) in dry THF were mixed and stirred for fifteen minutes at -10°C . A 0.5 M solution of cyclobutanone **3** (1 equiv.) in dry THF was added dropwise to the resulting ylide solution at -78°C . The mixture was stirred at room temperature for 3–5 hours. Addition of water and extraction with diethylether gave a crude product which was purified by flash chromatography on silica gel (AcOEt:cyclohexane, 2:8) to yield the two diastereoisomeric epoxides. Compound *endo*-**5a**: ^1H NMR (200 MHz, CDCl_3) δ 7.61 (d, 2H, $^3J=8.3$ Hz), 7.22 (d, 2H, $^3J=8.3$ Hz), 4.61 (dd, 1H, $^3J=8.3$ Hz, $^4J=1.4$ Hz), 2.91 (s, 3H), 2.70–2.58 (m, 1H), 2.55 (d, 1H, $^2J=4.6$ Hz), 2.36 (d, 1H, $^2J=4.6$ Hz), 2.42 (s, 3H), 2.25 (m, 1H), 1.70–0.70 (m, 8H); ^{13}C NMR (50 MHz, CDCl_3) δ 143.2, 136.7, 129.5, 126.9, 66.5, 57.5, 46.8, 35.6, 33.4, 31.4, 24.9, 24.1, 22.3, 21.5, 21.4; IR 2927, 2854, 1340, 1158; MS (EI) 322 (2%), 238 (24%), 185 (24%), 155 (100%), 91 (72%); Anal. calculated for $\text{C}_{17}\text{H}_{23}\text{NO}_3\text{S}$ C%: 63.52; H%: 7.21; N%: 4.36; S%: 9.97; Found: C%: 63.43; H%: 7.30; N%: 4.13; S%: 10.04. Compound *exo*-**5a**: ^1H NMR (200 MHz, CDCl_3) δ 7.70 (d, 2H, $^3J=8.2$ Hz), 7.28 (d, 2H, $^3J=8.2$ Hz), 4.83 (d, 1H, $^3J=9.1$ Hz), 2.79 (s, 3H), 2.72 (d, 1H, $^2J=9.0$ Hz), 2.49 (d, 1H, $^2J=9.0$ Hz), 2.42 (s, 3H), 2.38–2.26 (m, 2H), 1.90–0.90 (m, 8H); ^{13}C NMR (50 MHz, CDCl_3) δ 143.3, 136.4, 129.4, 127.2, 63.1, 57.9, 50.1, 32.9, 30.4, 29.6, 23.2, 22.4, 21.3, 21.4; IR 2929, 2852, 1340, 1161; MS (EI) 321 (4%), 238 (72%), 166 (36%), 155 (32%), 91 (60%), 42 (100%); Anal. calculated for $\text{C}_{17}\text{H}_{23}\text{NO}_3\text{S}$ C%: 63.52; H%: 7.21; N%: 4.36; S%: 9.97; Found: C%: 63.46; H%: 7.25; N%: 4.29; S%: 10.35.
- Feneau-Dupont J.; Declercq, J.-P.; Vanwetswinkel, S.; Genicot, C. *Acta Cryst.* **1993**, *49*, 561–562.
- General procedure for the rearrangement of **5**: lithium iodide (1 equiv.) was added to a 0.1 M solution of the mixture of epoxides **6** (1 equiv.) in dry THF. The resulting solution was refluxed for two hours, then cooled to room temperature. The crude product was purified by flash chromatography on silica gel (AcOEt:cyclohexane, 2:8). Compound **4a**: ^1H NMR (200 MHz, CDCl_3) δ 7.79 (d, 2H, $^3J=8.0$ Hz), 7.31 (d, 2H, $^3J=8.0$ Hz), 4.68 (d, 1H, $^3J=12.6$ Hz), 2.60 (s, 3H), 2.43 (s, 3H), 2.40–2.15 (m, 4H), 2.10–1.00 (m, 8H); ^{13}C NMR (50 MHz, CDCl_3) δ 213.5, 143.2, 136.8, 129.4, 127.6, 63.7, 43.9, 36.7, 31.4, 30.0, 29.2, 25.0, 24.2, 21.4, 19.6; IR 2927, 2854, 1749, 1338, 1156; MS (EI) 321 (28%), 165 (100%), 136 (100%), 91 (56%); Anal. calculated for $\text{C}_{17}\text{H}_{23}\text{NO}_3\text{S}$ C%: 63.52; H%: 7.21; N%: 4.36; S%: 9.97; Found: C%: 63.27; H%: 7.16; N%: 4.22; S%: 10.14. Compound **6a**: identical with an authentic sample: RN=81255-91-6.